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Colligative Properties of Ideal Dilute Solution – A useful way to determine the activity coefficient  
– Example 9.11: in 500 g of water, 24 g of a nonvolatile solute of MW 241 g/mol is dissolved. The observed freezing point depression is 0.359 °C. Calculate the activity coefficient of the solute.

### CHAPTER 9 IDEAL AND REAL SOLUTIONS - chen.chemistry.ucsc.edu

Ideal solutions: Non-ideal solutions: Positive deviation from Raoult's law: Negative deviation from Raoult's law: 1. Obey Raoult's law at every range of concentration. 1. Do not obey Raoult's law. 1. Do not obey Raoult's law. 2. neither heat is evolved nor absorbed during dissolution. 2. Endothermic dissolution; heat is absorbed. 2.

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### Relieving and Housebreaking - gdb-official.com

Solution Thermodynamics CH2351 Chemical Engineering Thermodynamics II Unit – I, II www.msubbu.in Dr. M. Subramanian Associate Professor Department of Chemical Engineering Sri Sivasubramaniya Nadar College of Engineering Kalavakkam – 603 110, Kanchipuram (Dist) ... solution. (ideal solution)

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### **Ideal Gas Law Problems - mmsphyschem.com**

A 1-molal solution contains one mole of solute per 1 kg of solvent. Molality is a hybrid concentration unit, retaining the convenience of mole measure for the solute, but expressing it in

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