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The first law of thermodynamics is a version of the law of conservation of energy, adapted for thermodynamic systems. The law of conservation of energy states that the total energy of an isolated system is constant; energy can be transformed from one form to another, but can be neither created nor destroyed. The first law is often formulated as $\Delta U = Q - W$. It states that the change in the internal ...

First law of thermodynamics - Wikipedia

1: Offered jointly with the School of Biological Sciences and The Paul Merage School of Business. See School of Biological Sciences section of the Catalogue for information.. 2: Offered jointly with the Donald Bren School of Information and Computer Sciences.

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