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5) Combustion Reaction – A chemical change where an element or a compound reacts with oxygen – This will often yield energy in the form of heat and light – Oxygen is ALWAYS going to be on the reactant side – This is the main indicator that a combustion reaction has occurred

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4.1 Writing and Balancing Chemical Equations By the end of this section, you will be able to: • Derive chemical equations from narrative descriptions of chemical reactions. • Write and balance chemical equations in molecular, total ionic, and net ionic formats.

4.1 Writing and Balancing Chemical Equations

4.1 Chemical Reactions and Chemical Equations 127 A chemical equation is a shorthand description of a chemical reaction. The following equation describes the burning of hydrogen gas to form liquid water.

Chapter a I to Chemical reactions - Mark Bishop

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electricity. Chemical reactions involve rearrangement of the atoms and produce new substances in this process. For example, in the above picture, just the mixing of two substances can cause the chemical reaction to occur. Chemical Reactions are everywhere in nature. The process by which the plants make their own food is called photosynthesis.

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