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To find the order for Mg, setup the equation like above so the concentration of oxygen cancels out. To find the order for O<sub>2</sub>, simply try to cancel the concentration of Mg.; For the rate constant, once you have the rate law, plug in any row of data from the chart and solve for k.

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### **7. A gas occupies a volume of 375 mL at 45°C. What will**

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