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In thermodynamics, a critical point (or critical state) is the end point of a phase equilibrium curve. The most prominent example is the liquid-vapor critical point, the end point of the pressure-temperature curve that designates conditions under which a liquid and its vapor can coexist. At higher temperatures, the gas cannot be liquefied by pressure alone.

**Critical point (thermodynamics) - Wikipedia**

Thermodynamics is the branch of physics that has to do with heat and temperature and their relation to energy and work. The behavior of these quantities is governed by the four laws of thermodynamics, irrespective of the composition or specific properties of the material or system in question. The laws of thermodynamics are explained in terms of microscopic constituents by statistical mechanics.

**Thermodynamics - Wikipedia**

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Thermo 1 (MEP 261) Thermodynamics An Engineering Approach Yunus A. Cengel & Michael A. Boles 7th Edition, McGraw-Hill Companies, ISBN-978-0-07-352932-5, 2008 Sheet 1: Chapter 1 "What is the difference between kg-mass and kg force? Solution

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More information: "Fundamentals of Engineering Thermodynamics, 8th Edition" by Yunus A. Cengel, Michael A. Boles, 2014, Wiley, ISBN: 978-1-119-28128-3 - m.blog.naver.com

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